

# Omenia

## Harrison Property

600264

## Deer Horn Mines Ltd

### Location

The properties is located on the eastern margin of the west range mountains just east of Tweedsmuir Park. It lies just north of Lindquist lake at an elevation of 4000 ft. (Map 1)

### Previous Work

In 1943 scheelite was found in slide material on the north shore of Lindquist lake by the Harrison brothers. In 1944 Frank Toubin discovered gold bearing quartz veins in the same area. Pioneer Gold Mines B.C. optioned the property in 1944 and worked on it until 1946.

Deer Horn Mines Ltd was incorporated in 1950 and bought the Harrison claims in 1950. They worked on the property in 1954 & 56.

### Property

The property consists of thirty crown granted claims and fractional claims, and 8 claims held by record. (Map 2)

General Geology

The property is on the eastern range border of the coast range mountains and at the eastern edge of the coast range batholithic complex. Granitic rocks underlie the southern half of the property while the northern half is underlain by sediments of the Hazelton group.

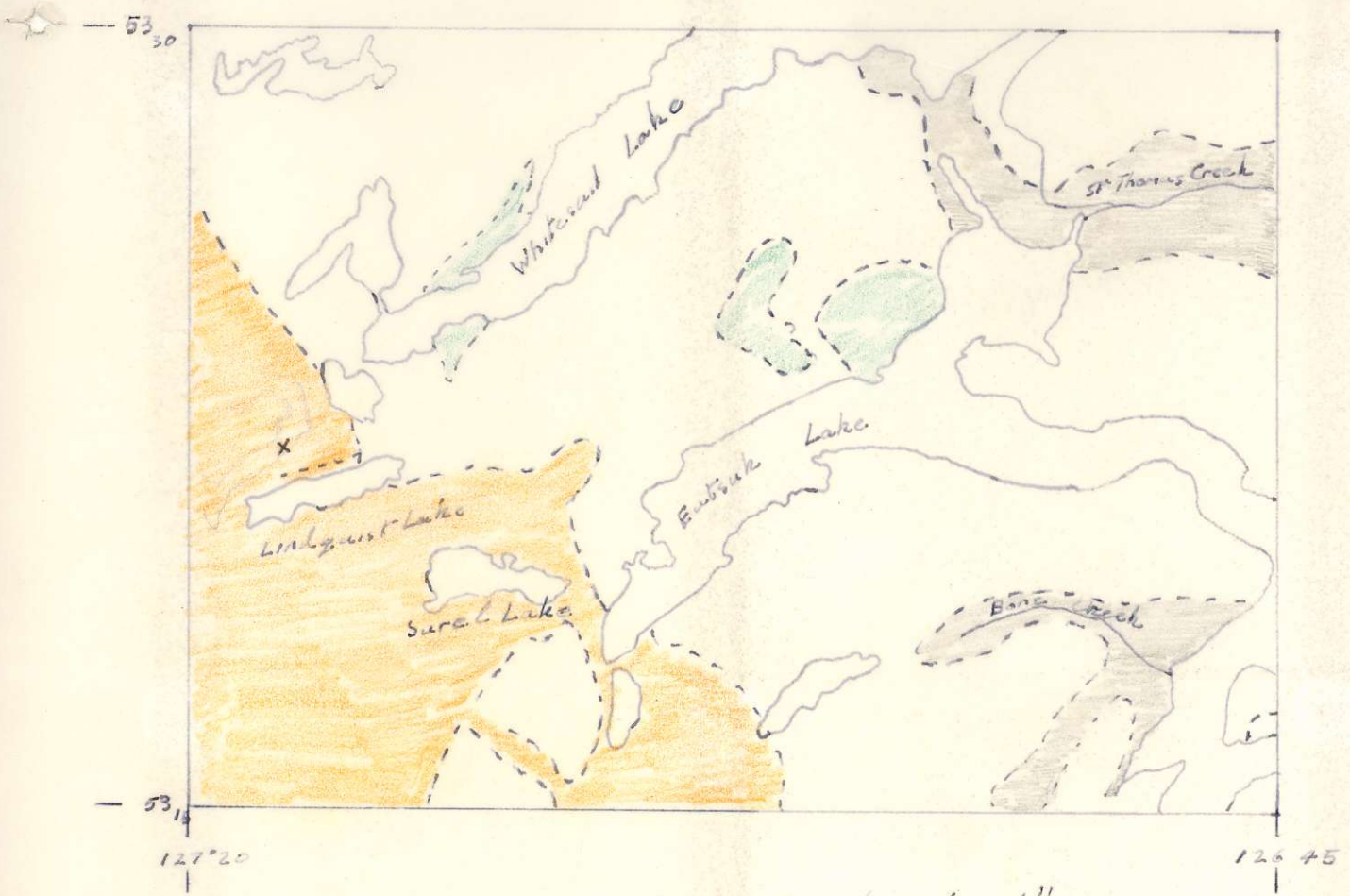
The granitic rocks intrusive into the sediments consist mostly of a pinkish weathering granite and minor amounts of medium grained quartz diorite.

The Hazelton sediments locally consist of slates sandy and silty gneisses argillite and tuff. Near the intrusive contact with the granitic rocks the sediments have been converted into andalusite schists.

Intense shearing and alteration has occurred in the vicinity of the contact. In places silicification extends 400ft into the sediments from the contact.

Mineralization occurs in quartz veins and in stringers associated with the silicification. The quartz veins appear to represent one faulted segment of one vein, or branches of a single vein structure. The apparent length of this main vein is 2600ft of which 1220ft has been explored; it dips at 30° to the north with a dip length of less than 300ft. Scheelite slides also occur on the property; the scheelite is apparently derived from otherwise barren quartz veins and stringers in rocks of the Hazelton group.

Location Map



Scale 3 miles to 1"



Recent Alluvium



Diorite



Coast range Batholith Granite diorite & granodiorite

Jurassic

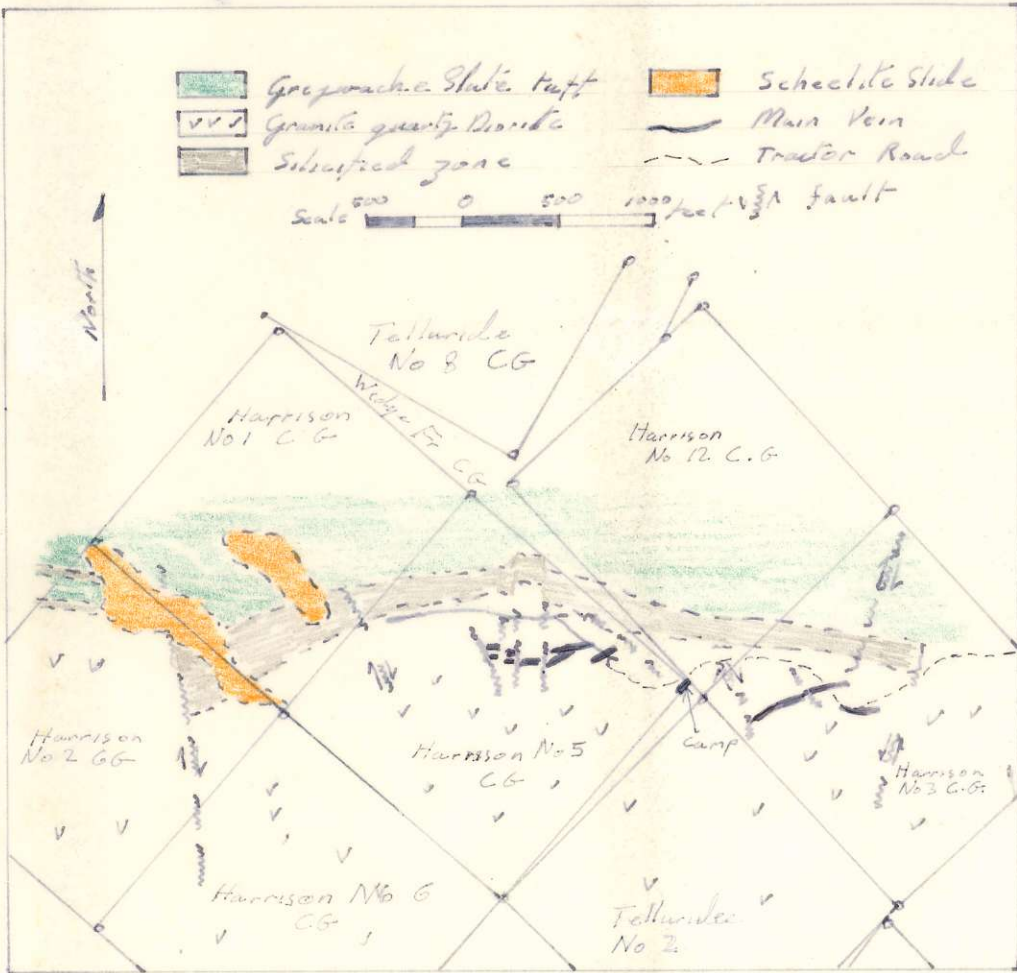


Hazellon group fragmental volcanics and sediments

x Location of property

HARRISON CLAIM GROUP

and Geology



Macroscopic Description

Hand Samples

Hand samples consist of sulphides to the extent of about 20% scattered through a milky matrix vein quartz matrix. Sphalerite occurs sparingly through the quartz matrix. The only opaque mineral apparent is quartz. Sulphides identifiable consist of pyrite, chalcopyrite, sphalerite and tellurides and galena. The sulphides occur as small masses scattered through the quartz matrix; pyrite with associated chalcopyrite and sphalerite occurs as masses up to 3cm in diameter in the quartz wall while the tellurides fill small joint fractures or occur as pockets generally less than 1cm in diameter. vugs do occur in the quartz but it is not clear whether these represent unmineralized spaces or places where tellurides have been weathered out.

Approximate Relative proportions of sulphides

- Pyrite 50%
- Chalcopyrite 10%
- Sphalerite 5%
- Galena 5%
- Tellurides 30%

Microscopic Description

Minerals identified with %s

	Pyrite	40%	
	Chalcopyrite	20%	
	Sphalerite	15%	
	Arsenopyrite	< 3%	
	Galena	< 5%	
	Chalcoite	< 3%	
	Scheelite	< 3%	
	Pyrrhotite	< 3%	
Tellurides	Tellurobismuth	} Together < 15%	with Weholite
	Tetradymite		
	alluaite	< 5%	
	Petzite	} Together < 10%	
Hemite			
	Weholite		
Tentatively identified	Erubite	< 1%	
	Stannite	< 1%	
	Gold	< 1%	
Unnamed probably	Tellurides		
	Mineral A	< 1%	
	Mineral B	< 1%	

Mineral %s are approximate as it is unrealistic to give specific figures when dealing with over 20 polished sections

The B<sub>1</sub> tellurides are considered together as are Petzite and Hemite as it is difficult to estimate their relative proportions.

Minerals      Characteristic Microscope Properties

Pyrite - yellow, hard with a poor polish  
isotropic in crossed nicols.

Chalcopyrite - yellow moderate polish and  
moderately hard, isotropic.  
succ Eteck tests -ve to KCN

Sphalerite - grey low reflectivity  
medium hardness isotropic  
but with red internal reflection

arsenopyrite - whiter than pyrite and  
anisotropic.  
Eteck tests H<sub>2</sub>Cl<sub>2</sub> stains brown  
stain washed off

Tellurobismuth - PPL, white and soft  
with high reflectivity. CN good  
anisotropic  
Eteck tests

NH<sub>4</sub>O<sub>2</sub> +ve effervesces and quickly  
stains black

FeCl<sub>3</sub> +ve quickly stains iridescent  
to black.

KOH weak +ve

KCN -ve

Hg. Cl<sub>2</sub> -ve positive stains brown

Tetradymite - PPL very pale pink high  
 reflectivity greater than 50%  
 CN anisotropic Hardness  
 low

Etch Tests

$NH_4O_3$  weak positive or negative  
 $FeCl_3$  positive but not as strong  
 as for Tellurides  
 $KCN$  negative  
 $HgCl_2$  negative  
 $KOH$  negative

Galena - PPL white soft with high  
 reflectivity  
 CN isotropic complete extinction

Etch Tests

$FeCl_3$  stains green and iridescent  
 $KOH$  negative  
 $KCN$  negative  
 $HgCl_2$  negative  
 $HNO_3$  stains brown <sup>no</sup> effervescence  
 $HCl$  stains brown to iridescent  
 slowly.

Heulandite - PPL grey moderate to low  
 reflectivity hardness low  
 CN strong anisotropic blues  
 and browns.

Etch Tests

$FeCl_3$  positive  
 $HgCl_2$  positive  
 $NH_4O_3$  positive weak  
 $KCN$  positive  
 $KOH$  negative



Schedelite - identified mainly on the basis of fluorescence.

Optical properties - low reflectivity high internal reflection, softer than quartz.

Wehrlite - white to pale grey react to Telluric acid, high reflectivity moderate anisotropism

Fluor tests

NH<sub>4</sub>O<sub>3</sub> +ve stains brown

FeCl<sub>3</sub> stains brown quickly

KOH negative

H<sub>2</sub>O negative

HCl negative

KCN negative

Allite - white isotropic but does not show complete extinction reflectivity high, hardness low.

Fluor tests

NH<sub>4</sub>O<sub>3</sub> positive and effervesces

FeCl<sub>3</sub> positive

HCl positive

KCN negative

H<sub>2</sub>O "

KOH "

Chalcoite - Grey blue with low reflectivity and hardness isotropic

Stannite ? Much browner than pyrochroite same reflectivity and hardness anisotropic occurred as a small grain associated with pyrochroite etch tests could not be performed

Mineral B Grey with low reflectivity and hardness isotropic  
HNO3 positive no effervescence  
KCN negative  
FeCl3 negative  
H2Cl negative  
KOH negative

Mineral A Very pale grey pink slightly darker than Tellurobismuth hardness and reflectivity similar to Tellurobismuth.  
Isotropic to weakly anisotropic  
Etch tests  
HNO3 stains brown  
FeCl3 positive not as strong as for other tellurides  
KCN positive woodcock stain  
KOH negative  
HCl negative  
H2Cl negative or weakly positive with a brown stain

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Scheelite

Scheelite mineralization is characterized  
strictly associated with acid intrusions  
as it is in this case. Tungsten probably  
contained in a late fluid phase  
differentiated from the magma permeates  
the surrounding country rock, usually  
being deposited as scheelite. Generally  
the calcium required for this is derived  
from the sediments. Thus many deposits  
of calcium scheelite are found in calcium  
rich rocks. In this area it is  
not clear where the necessary calcium  
came from, the luffs could be a  
likely source.

Contact metamorphism of the  
sediments adjoining the intrusion is of  
the hornblende hornfels facies, this  
indicates a temperature of  $500-450^{\circ}\text{C}$   
fairly high for this type of intrusion.  
Thus if the scheelite is restricted to  
rocks of this grade of metamorphism the  
extent of scheelite mineralization is probably  
minor.

The relationship of scheelite to  
sulphides is not clear. The BC Ministry  
of Mines report 55 mentions that scheelite  
is in otherwise barren quartz veins. This  
suggests that the sulphides were carried  
past the area of scheelite deposition  
to a place where a lower geothermal  
gradient allowed them to crystallize. This  
of course suggests that the Tungsten  
sulphide and telluride mineralizations  
are all associated with the one intrusion.

Textures of the early sulphides

Included under under this heading are pyrite sphalerite chalcopyrite galena and pyrrhotite also arsenopyrite and stannite?

Much of the pyrite sphalerite and chalcopyrite seen in the polished sections occurred separately from the tellurides. Pyrite generally occurred as corroded fractured grains with pitted surface. In some places the pyrite was anisotropic but yellow colour and hardness were still sufficiently characteristic. Associated with the pyrite some grains of equal hardness but much lighter colour may represent arsenopyrite. Generally the corroded appearance of the pyrite was directly attributable to replacement by sphalerite which formed a ground mass for the fractured pyrite crystals. In other places the corroded pyrite crystals were enclosed in the quartz matrix. Occasional inclusions of chalcopyrite were seen in the pyrite but not as frequently as the replacing sphalerite. Evidence for replacement of pyrite by sphalerite was considered to be atolla texture and non matching vein walls of sphalerite in pyrite.

Where chalcopyrite formed the ground mass to the pyrite (Fig. 11) there was considerably less evidence for replacement (Figure 12) and generally the chalcopyrite seems to have filled fractures in the pyrite crystals by displacing the broken parts. Occasional isohedral blebs of sphalerite do occur in the chalcopyrite; these blebs often have inclusion texture. Small blebs of galena also occurred associated with the chalcopyrite.

Fine Emulsion texture is common in the sphalerite though in places % of chalcopyrite and its in sphalerite and its fingerling shape indicates replacement of sphalerite by chalcopyrite (Fig 3)

Running through the chalcopyrite or sphalerite ground mass at fine veins of chalcocite were observed (Fig 1) these either formed an an echelon pattern down the centre of sphalerite or chalcopyrite veins, or radiated out from perpendicularly from pyrite grains. The extremely fine nature made identification uncertain but they appear to represent secondary material though but replacement has not progressed to any extent.

Pyrite was nowhere observed in contact with the tellurides. Sphalerite and chalcopyrite where in contact with the tellurides exhibit mutual boundary texture except in one place tellurides vein into chalcopyrite without any apparent replacement.

Pyrohalite occurs as rounded grains contained in the tellurides or along telluride chalcopyrite contacts (Figure 7) In places separated grains of pyrohalite contained in the tellurides do remain in optical continuity indicating that the tellurides have replaced the pyrohalite (Fig 7). In another place a grain of pyrohalite apparently centrally replaced by tellurides contains 2 separated parts of a grain of stannite? which are still in optical continuity; this again suggests replacement (Figure 4). The stannite could not be identified with any certainty because only one small grain was seen.

Atoll Texture in Pyrite crystals

Core replacement of pyrite by sphalerite is fairly common; chalcopyrite nor any other mineral replaces pyrite to any extent.

Cores of sphalerite are generally joined to the ground mass sphalerite by a fine vein of sphalerite. In some cases pyrite appears to contain an isolated grain of sphalerite but this is probably just an effect of the angle of the section. Replacement of pyrite also operated around the edges of the pyrite crystals, giving them their present corroded shape. Core replacement then acted as well as rim replacement and not in preference to it. As there is no way of knowing the extent of rim replacement the core replacement may have been very minor, now over emphasized by a greater amount of rim replacement.

Emulsion Texture in Sphalerite

This takes the form of fine arched blebs of chalcopyrite randomly distributed through sphalerite. Paths of chalcopyrite orientated along cleavage directions in the sphalerite do not constitute emulsion texture but will be considered here.

The texture most likely originates by exsolution of chalcopyrite from sphalerite at temperatures generally above 300°C. Up to 10% chalcopyrite can be held in solution so that in instances where more than 10% chalcopyrite is found in sphalerite (Fig 3) a process other than exsolution has been involved. If the blebs originate by exsolution their size and distribution depends on ① the % of chalcopyrite in solution ② the cooling history.

As sphalerite found in pyrite (forming a replacement atal texture) contains no chalcopyrite it appears that at the temperature at which replacement proceeded the solubility of chalcopyrite in sphalerite must have been very low. In the case of slow continued growth over a temperature range, 2 factors would suggest a higher % of chalcopyrite in the centre of crystals of sphalerite; that is ① it crystallized at higher temperatures and therefore greater solubility of chalcopyrite in sphalerite.

② it has had more time to exsolve chalcopyrite. If on the other hand the sphalerite crystal is held at a temperature suitable for diffusion chalcopyrite would migrate to crystal boundaries, and thus blebs of chalcopyrite would be larger near the sphalerite crystal boundaries. It is noticeable in this suite that in sphalerite associated with

the tellurides the chalcopyrite blebs are larger near the sphalerite crystal boundaries (Fig 2) while in sphalerite associated with pyrite and chalcopyrite if any gradation is seen it is the reverse. In one case where sphalerite and chalcopyrite were associated with tellurides the sphalerite chalcopyrite contact was extremely uneven this could easily be formed by chalcopyrite blebs arriving at the boundary (Fig 2) The possibility of draining of chalcopyrite from the borders of sphalerite crystals where they adjoin chalcopyrite crystals seems unlikely, in the case where the sphalerite chalcopyrite boundary is smooth yet chalcopyrite blebs remain in the centre of the sphalerite crystal. A more plausible explanation seems to be that given for the case of smaller chalcopyrite blebs near the sphalerite crystal boundary.

The orientated laths of chalcopyrite in sphalerite are much larger than the embulsion texture chalcopyrite but occur in association with it. If these laths originated for by diffusion it seems unlikely that the smaller blebs would survive in their vicinity; thus it appears that most of the orientated laths of chalcopyrite are of replacement origin. This is supported by the thickening of laths where two intersect.

A reasonable explanation for these textures seems to be that sphalerite associated with pyrite has undergone little post crystallization diffusion. This is also suggested by the fact that smaller sphalerite grains in this association have finer embulsion textures. This is despite replacement of sphalerite by chalcopyrite. Sphalerite associated with the tellurides appears to have undergone more



post crystallization diffusion possibly caused  
by the later crystallization of the tellurides.

Tellurides - Textures and general Description

These include Tellurites, Tetradymite, Hemite, Petzite, Altaitite identified with less certainty was ~~was~~ mercurite also considered here is Cosalite and the two unnamed minerals.

Tetradymite and Tellurites - these two minerals like the other tellurides occupy small areas in the quartz matrix rarely exceeding 5mm in diameter. Often the quartz telluride contact exhibits a pseudo cavity texture of quartz "biting" into the tellurides nowhere are any embayment quartz crystal boundaries outlined against tellurides. Replacement of tellurides by quartz at the temperatures at which the tellurides would be deposited (<200°C) is extremely unlikely conversely telluride quartz contacts do not suggest that the tellurides have replaced the quartz. Thus it appears that the tellurides filled pre-existing spaces, in which case the shape of these cavities suggests that the quartz originally crystallized as a gel which contracted to form cusped cavities and fine fractures.

Generally these two tellurides occur as a fine intergrowth in about equal parts. When in contact the tellurites appears very slightly pinker otherwise physical properties are the same. The intergrowth though just apparent under crossed nicols can be emphasized by etching with either a diluted solution of FeCl<sub>3</sub> or NH<sub>4</sub>O<sub>3</sub> conc between 1/1 - 1/2. Tellurites and tetra tetradymite are in about equal proportions; the proportions of the two minerals does not appear to vary

to any great extent. Generally the laths are crystallographically controlled though in a few places a finer symplectitic type of intergrowth was observed. As separated laths were in optical continuity and appeared to make up parts of a whole crystal the intergrowth is post crystallization. Tetradymite and Tellurobismuth have the same crystal symmetry with unit cells nearly the same size. Thus it is possible for a compound consisting of a mixture of the two minerals to crystallize and later exsolve into the two distinct minerals. The uniformity of the intergrowths throughout the tellurides general crystallographic control of the laths and the mutual boundary relationships between the laths rules out a replacement origin for the intergrowth.

Mineral A occurred sparingly within these two tellurides it was slightly pinker and darker than the Tellurobismuth. As the laths of Tellurobismuth and tetradymite stone abruptly set it against this mineral it must represent an early exsolved mineral or a mineral trapped in the tellurides during crystallization. The association with properties and optical properties suggest that it is probably another telluride.

Pelzite and kessite are closely associated with these two tellurides. Pelzite in some places forms a symplectitic intergrowth in these two tetradymite and Tellurobismuth near the crystal contacts. The restriction of this intergrowth to crystal boundaries suggests replacement (Fig 8) though if this is the case the pelzite has replaced both

tellurides with equal vigour. If this is not the case the petzite must have been exsolved from the tetradymite tellurobismuth solid solution before they exsolved. In one polished section petzite which replaces tetradymite and tellurobismuth is itself being replaced by Hermitite (Fig 6). The ~~curate~~ contact of petzite with the tellurides suggests replacement rather than exsolution. Hermitite appears to have replaced petzite where it fingers into the tellurides leaving small areas of petzite at the end of the fingers. This hermitite shows a mottled anisotropism suggesting that it crystallized above 149°C as the ~~was~~ disordered form.

Hermitite also exhibits asynchroitic intergrowth in <sup>occurs</sup> allatite apparently as a replacement rather than exsolution.

In one section a white bismuth telluride with optical and etch properties similar to tellurobismuth was found. The colour was whiter than that of tellurobismuth suggesting that this was wehrlite. On etching this mineral with diluted  $FeCl_3$  an extremely fine pinedo cutectic intergrowth became apparent. This intergrowth could not be seen on polished surfaces as a change in colour or relief. It is not clear whether this intergrowth represents 2 phases of wehrlite or wehrlite and another mineral. Five needles 20 microns long of tetradymite cut across this intergrowth in wehrlite as there was no thickening where two needles intersected the tetradymite is considered to have been exsolved from the wehrlite before the formation of the intergrowth. Tetradymite which crystallized separately from the wehrlite but in contact with it appears to have been

replaced by the ueholite to a small extent.  
(Figure 10).

altaitite occurs scattered through the quartz as small anhedral blebs. This telluride was distinguished from galena by ① altaitite does not have complete extinction ② altaitite has stronger reaction to  $NH_4O_3$  and generally effervesces. On the basis of these facts some of the white blebs scattered through the quartz represent galena; occasional occurrence of triangular cleavage pits confirmed this. Generally the altaitite occurred separately from the other tellurides ~~from~~ or in contact with kersite.

Coralite identified as such, by white colour anisotropism and positive KOH test occurred as very fine blebs in chalcocyanite (10 microns).

Gold - occasional blebs were seen in the ueholite these were extremely small (less than 10 microns)

Mineral B a grey exsolution mineral in Tellurobismuth had etch properties which appeared to rule out Kersite or Petzite as possibilities that is mainly a negative reaction to  $FeCl_3$ . The mineral occurred as fine laths which ended abruptly against a Tellurobismuth Tellurobismuth crystal contact (Figure 5)

## Paragenesis

Mineralization appears to be associated with the intrusion; it is not clear from the hand specimens and polished sections whether there was any major break in mineralization between firstly the deposition of the scheelite and sulphides and secondly between the deposition of sulphides and tellurides.

The importance of shearing mentioned in The BC Ministry of Mines report 1955 is unknown; only the pyrite shows minor effects of stress in the form of fractures; the gangue appears to be massive and undeformed.

Scheelite represents the earliest phase of mineralization probably related to the contact metamorphism of the adjacent sediments. The silica forming the silicified zone and veins, in which the scheelite was deposited is probably mostly of igneous origin though some could have been sweated out of the sediments.

The relationship of scheelite to the pyrite, sphalerite and chalcopyrite is unknown. But these minerals with the other sulphides were probably deposited next, though after the country rock had cooled to about 400 to 350°C, or at such a distance from the contact that the country rock was at this temperature. First of the sulphides to crystallize was pyrite; following minor fracturing sphalerite and chalcopyrite crystallized in the interstices<sup>es</sup> of the surrounding the pyrite grains.

General textures suggest simultaneous crystallization of the sphalerite and chalcopyrite; thus when the sphalerite crystallized chalcopyrite was available

to go into solid solution with it. as the % of dissolved chalcopyrite in the sphalerite is fairly high (near the maximum possible), the sphalerite probably crystallized in the upper part of its temperature range. The presence of chalcopyrite in solid solution in the sphalerite indicates that the chalcopyrite was not all removed from solution by crystallization, before the sphalerite started to crystallize. The presence of sphalerite grains with emulsion texture contained in the chalcopyrite and making up more than 15% of the overall volume strongly suggests that the chalcopyrite continued to crystallize after the sphalerite. The textural evidence suggesting that chalcopyrite replaces sphalerite also supports this. Thus the sphalerite and chalcopyrite form an example of overlapping deposition. The replacement of chalcopyrite by sphalerite and the veining nature of chalcopyrite and sphalerite through pyrite indicates that these two minerals crystallized after the pyrite.

Arsenopyrite appears to have crystallized with the pyrite. Pyrrhotite was probably associated with the chalcopyrite crystallization.

The tellurides represent a temperature of crystallization considerably lower than that of the sulphides. The close association of the tellurides and sulphides suggests 3 possibilities: (1) the tellurides were deposited some time after the sulphides when the country rock had cooled sufficiently to permit them to crystallize; (2) that the country rock was already below the the temperature of crystallization of

The tellurides and the fall in temperature and change in composition of the ore fluid caused the tellurides to deposit; (3) the tellurides represent a separate mineralization of the core rock. The second possibility appears most plausible. If the first possibility had occurred then one would expect considerable diffusion of chalcopyrite out of sphalerite bearing fairly coarse blebs and probably a leached zone in the sphalerite near chalcopyrite sphalerite contacts. This is not seen. Had the tellurides represented a separate mineralization one would expect to see some evidence of fracturing of the chalcopyrite and sphalerite caused by renewed heating; and veining of tellurides in these minerals. This is not seen.

The order of crystallization of the tellurides is not clear. The bismuth tellurides probably crystallized as solid solutions of two or more minerals probably at about the same time. Hercite and Petzite appear to be a little later with berrite the later of the two. Allrite probably crystallized at the same time as the bismuth tellurides.

Not mentioned above the galena and covellite probably crystallized in association with the chalcopyrite.

The gold ore is closely associated with the bismuth tellurides.

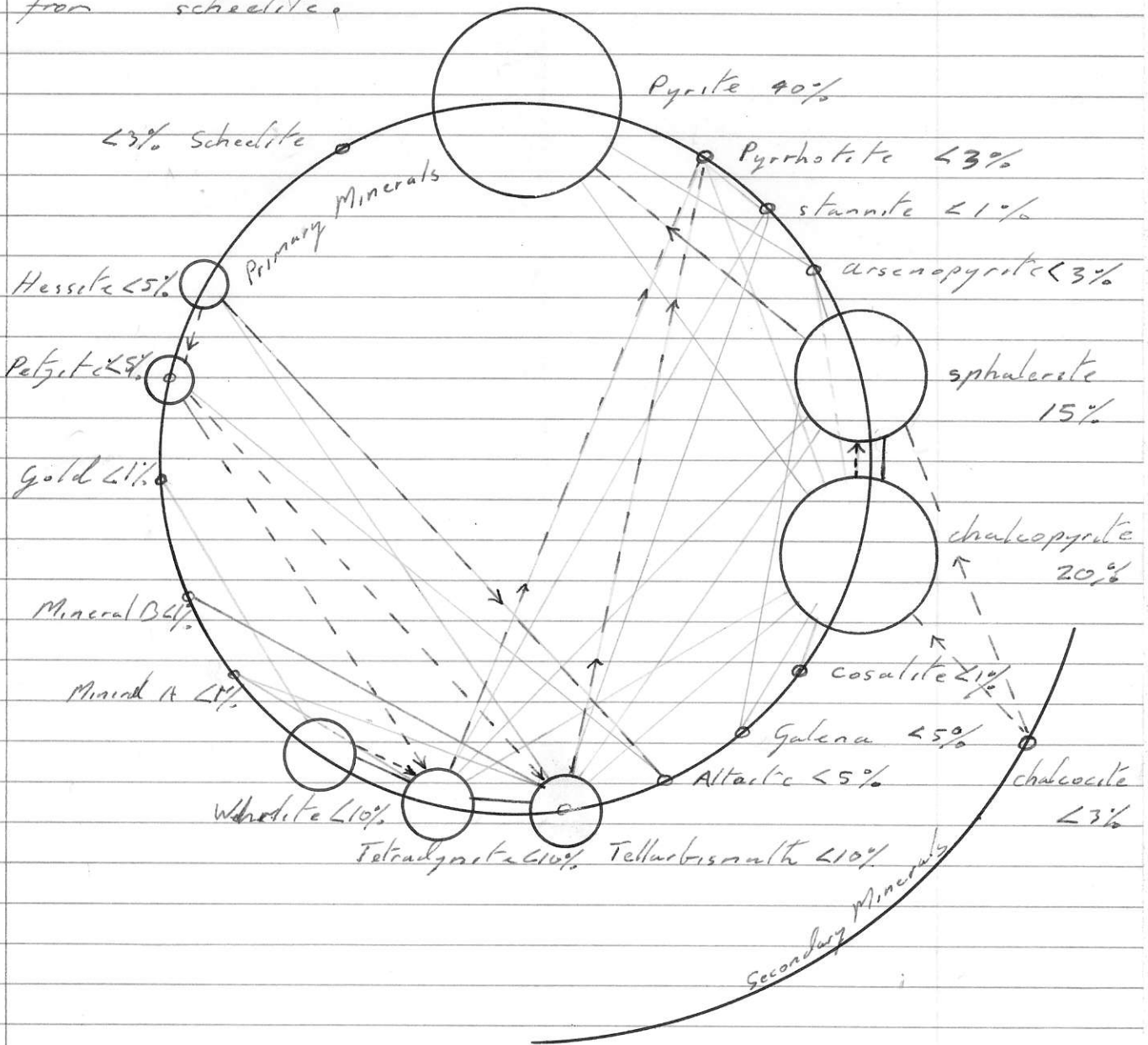


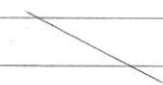
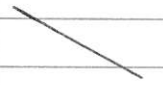
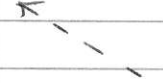
Classification of Deposit

This is a hydrothermal vein deposit.  
of the scheelite is included in the ore minerals.  
mineralization extended over the temperature  
range 450°C - 150°C at least.

Paragenesis

Approximate depositional sequence clockwise from scheelite.



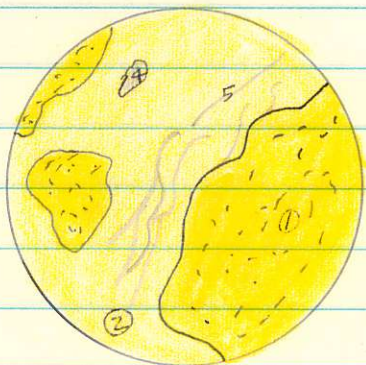
 joins minerals in contact  
 joins minerals forming exsolution intergrowths  
 joins minerals with replacement relationships arrow shows direction of replacement

### Diagrammatic representation of Paragenesis of Main Minerals



Temperatures indicated are only approximate guides.

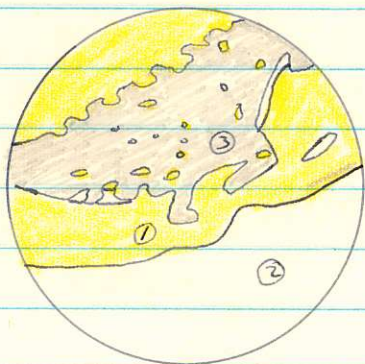
Figure 1



Pyrite ① with a ground mass of chalcopyrite ② occasional blebs of sphalerite ③ in the chalcopyrite and chalcopyrite trace. Fine veins of a secondary mineral ⑤ (Chalcocite?) occur in the chalcopyrite.

Section 9 High Power field of View .4mm

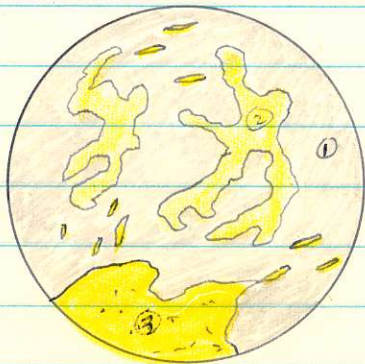
Figure 2



Chalcopyrite ① in contact with tellurides ② shows ventral boundary texture while the sphalerite ③ chalcopyrite boundary suggests replacement or diffusion of chalcopyrite to boundary by its serrate nature

Section T 3 High Power field of View .4mm

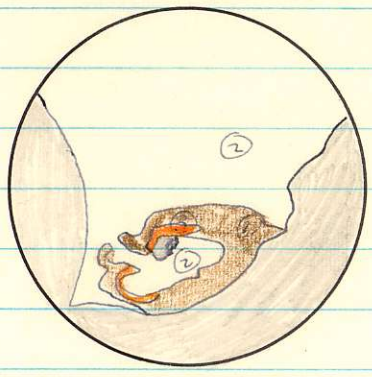
Figure 3



Matrix of sphalerite ① and chalcopyrite ② between pyrite fragments ③. chalcopyrite occurs as orientated laths and as anhedral fingerling masses suggestive of replacement.

Section 7A Medium Power field of view 1.5mm

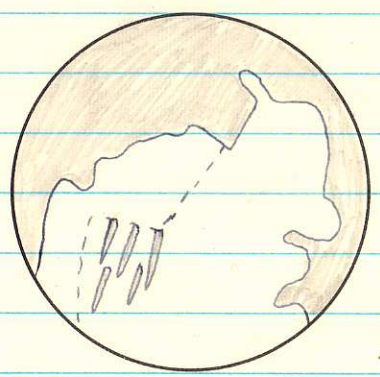
Figure 4



Pyrrhotite ① occurs as one grain  
 whose core has been  
 largely replaced by tellurides ②  
 2 parts of a stannite? chrysolite  
 still in optical unity occur in  
 the pyrrhotite. Grey in gangue  
 quartz.

Section T3 High Power field of view .4mm

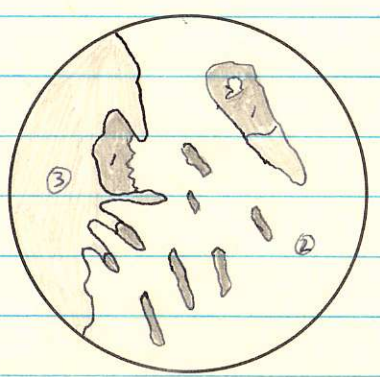
Figure 5



Tellurobismuth with fine  
 needle laths of grey mineral B  
 laths stop at crystal boundaries  
 suggesting exsolution.  
 Grey is quartz gangue

Section T1 High Power field of view .4mm

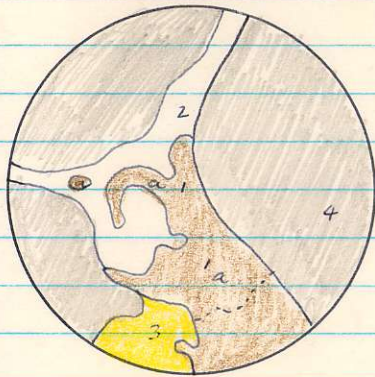
Figure 6



Petzite ① as a pseudo-embryo  
 in 2 B<sub>1</sub> tellurides being replaced  
 by hemite ③. B<sub>1</sub> tellurides  
 ② are exsolution intergrowth  
 of Tetradymite and Tellur  
 bismuth.

Section A High Power field of view .4mm

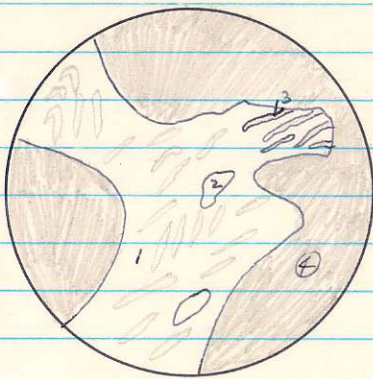
Figure 7



Pyrohalite ① being replaced by  
tellurides ② all Pyrohalite ①  
is in optical continuity. ③ is  
chalcopyrite. ④ is quartz gangue

Section T3 High Power field of view .4mm

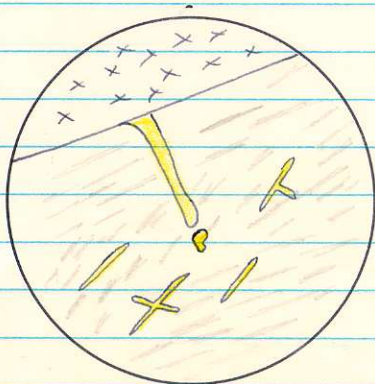
Figure 8



① Tellurides, Tetradyrite with  
baths of Tellurobasalt  
② is mineral A  
③ is superheated of pyrite  
④ Gangue of quartz

Section T3 High Power field of view .4mm

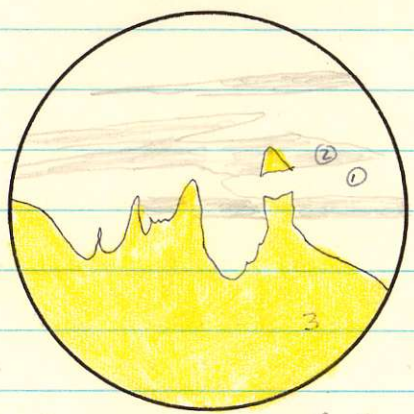
Figure 9



2 tellurides form a fine orient  
ated intergrowth one is probably  
wollastite  
An second the telluride Tetrady  
rite forms exsolution lamellae  
These lamellae show no thickening  
at intersections strongly suggesting  
exsolution rather than replace

Section T2 High Power view. The baths of this mined  
field of view .4mm was the earlier telluride baths  
so that it represent an earlier  
exsolution.

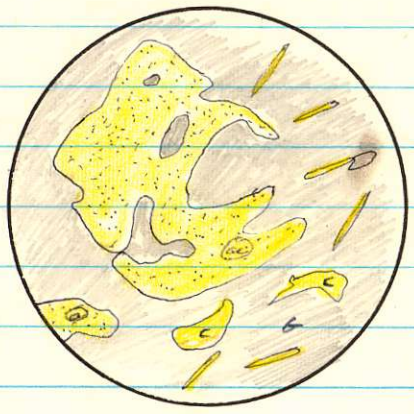
Figure 10



a pseudo eutectic of a white anisotropic mineral (1) and greyer anisotropic mineral (2) replacing a yellow mineral Tachygrite (3).  
 1 and 2 are Bi. Tellurides one or both being Wchelite.

Section T 2 High Power field of view .4 mm

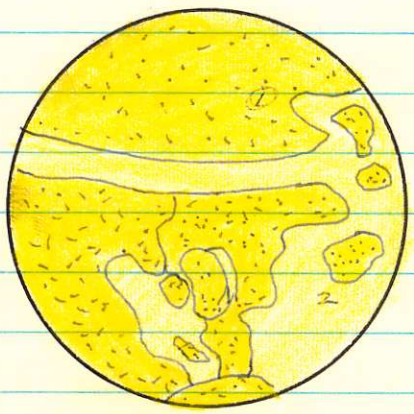
Figure 11



Pyrite being replaced by sphalerite with replacement working out from the centre of the pyrite crystals. Orientated lathes of chalcopyrite in the sphalerite.  
 @ Pyrite @ sphalerite  
 © chalcopyrite

Section 7A Low power Field of View 3mm

Figure 12



Pyrite (1) fractured grains surrounded by chalcopyrite (2) apparently with little or no replacement.

Section 9 Low Power field of view 3mm